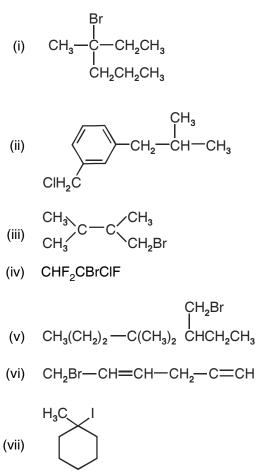
## Haloalkanes and Haloarenes

(Assignment - 1)

1. Write the IUPAC names of the following compounds.



(viii) CH<sub>3</sub>COCH<sub>2</sub>CHCICH<sub>3</sub>.

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(ix) 
$$CH_3 - CH - CH - CH - CH_3$$
  
 $HC - CH_3$   
 $H_3C$ 

- (x)  $(CCI_3)_3$  CCI
- 2. Write the structure of following halogen compounds
  - (i) DDT
  - (ii) Freon 12
  - (iii) 1–Bromo 4-sec-butyl-2-methylbenzene.
  - (iv) p-Bromotoluene.
  - (v) Methylene chloride
  - (vi) Chloral
- 3. Arrange the following in the increasing order of property indicated :
  - (i) Bromomethane, bromoform, chloromethane, dichloromethane. (Increasing order of boiling points).
  - (ii) 1-Chloropropane, isopropyl chloride, 1-chlorobutane (Increasing order of b.pt.)
  - (iii) Dichloromethane, chloroform, carbon tetrachloride. (Increasing order of dipole moment).
  - (iv) CH<sub>3</sub>F, CH<sub>3</sub>CI, CH<sub>3</sub>Br, CH<sub>3</sub>I (Increasing reactivity towards nucleophilic substitution).
  - (v) o, m, p-dichlorobenzenes (Increasing order of melting points).

$$\bigcirc CH = CH_2 + HBr \longrightarrow$$

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