PRACTICE TEST

Level 1

CLASS: XII

Unit 11: ALCOHOLS, PHENOLS AND ETHERS

Full marks: 20

Time: 40 Min

Eiwari

| Q.No | Questions | Μ |
|------|--|---|
| 1 | Write IUPAC name of (CH ₃) ₂ CH-OCH ₃ . | 1 |
| 2 | Which of the following is most acidic? | 1 |
| | (i) Benzyl alcohol (ii) Cyclohexanol (iii) Phenol (iv) <i>m</i> -Chlorophenol | |
| 3 | Arrange the following compounds in increasing order of boiling point: | 1 |
| | Propan-1-ol, butan-1-ol, butan-2-ol, pentan-1-ol | |
| 4 | What is the structure and IUPAC name of glycerol? | 1 |
| 5 | Out of <i>o</i> -nitrophenol and <i>p</i> -nitrophenol, which is more volatile? Explain. | 2 |
| 6 | Write the mechanism of hydration of ethene to yield ethanol. | 2 |
| 7 | You are given benzene, conc. H_2SO_4 and NaOH. Write the equations for the | 2 |
| | preparation of phenol using these reagents. | |
| 8 | Name the reagents used in the following reactions: | |
| | (i) Oxidation of a primary alcohol to carboxylic acid. | |
| | (ii) Oxidation of a primary alcohol to aldehyde. | |
| | (iii) Bromination of phenol to 2,4,6-tribromophenol. | |
| | (iv) Benzyl alcohol to benzoic acid. | 2 |
| 9 | How are the following conversions carried out? | |
| | (i) Propene \rightarrow Propan-2-ol. | |
| | (ii) Benzyl chloride \rightarrow Benzyl alcohol. | 2 |
| 10 | Explain the following with an example. | |
| | (i) Kolbe's reaction. | |
| | (ii) Reimer-Tiemann reaction. | |
| | (iii) Williamson ether synthesis. | 3 |
| 11 | Show how will you synthesise: | |
| | (i) 1-phenylethanol from a suitable alkene. | |
| | (ii) cyclohexylmethanol using an alkyl halide by an SN^2 reaction. | |
| | (iii) pentan-1-ol using a suitable alkyl halide? | 3 |
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